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STRUCTURE FILE UPDATES: 18 SEP 2007 HIGHEST RN 947490-11-1  
DICTIONARY FILE UPDATES: 18 SEP 2007 HIGHEST RN 947490-11-1

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=> d l75 ide can

L75 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2007 ACS on STN

RN 1314-36-9 REGISTRY

ED Entered STN: 16 Nov 1984

CN Yttrium oxide (Y2O3) (CA INDEX NAME)

OTHER NAMES:

CN BB

CN BB (metal oxide)

CN Diyttrium trioxide

CN FCD-660-2

CN Nanotek Y2O3

CN PC-YH

CN RU-PB

CN SY

CN SY (oxide)

CN Y-F

CN Y-F (metal oxide)

CN YO 3-245

CN Yttria

CN Yttrium oxide

CN Yttrium oxide (YO1.5)

CN Yttrium sesquioxide

CN Yttrium trioxide

CN Yttrium(3+) oxide

CN Yttrium(III) oxide

AR 11130-29-3

MF O3 Y2

CI COM, MAN

LC STN Files: AGRICOLA, ANABSTR, BIOSIS, CA, CAOLD, CAPLUS, CASREACT, CBNB,  
CHEMCATS, CHEMLIST, CIN, CSCHEM, CSNB, DETHERM\*, HSDB\*, IFICDB, IFIPAT,  
IFIUDB, IPA, MEDLINE, MSDS-OHS, PIRA, PROMT, RTECS\*, TOXCENTER, TULSA,  
USPAT2, USPATFULL, USPATOLD, VTB

(\*File contains numerically searchable property data)  
Other Sources: DSL\*\*, EINECS\*\*, TSCA\*\*  
(\*\*Enter CHEMLIST File for up-to-date regulatory information)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

\*\*PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT\*\*

42996 REFERENCES IN FILE CA (1907 TO DATE)  
999 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA  
43107 REFERENCES IN FILE CAPLUS (1907 TO DATE)  
55 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

REFERENCE 1: 147:290779  
REFERENCE 2: 147:290554  
REFERENCE 3: 147:290552  
REFERENCE 4: 147:290111  
REFERENCE 5: 147:290096  
REFERENCE 6: 147:289995  
REFERENCE 7: 147:289882  
REFERENCE 8: 147:289342  
REFERENCE 9: 147:288968  
REFERENCE 10: 147:287965

=> fil hcaplus

FILE 'HCAPLUS' ENTERED AT 15:45:58 ON 19 SEP 2007  
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FILE COVERS 1907 - 19 Sep 2007 VOL 147 ISS 13  
FILE LAST UPDATED: 18 Sep 2007 (20070918/ED)

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This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d 173 bib abs hitind hitstr retable tot

L73 ANSWER 1 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN  
 AN 2007:391759 HCAPLUS Full-text  
 TI The effects of process parameters on mechanical property and visual aspect of plasma sprayed Y2O3 coating  
 AU Seok, Hyun Kwang; Lee, Jae Gun; Baik, Kyeong Ho  
 CS Advanced Metals Research Center(AMRC), Korea Institute of Science and Technology, Seoul, 136-791, S. Korea  
 SO Materials Science Forum (2007), 539-543(Pt. 2, THERMEC 2006), 1212-1217  
 CODEN: MSFOEP; ISSN: 0255-5476  
 PB Trans Tech Publications Ltd.  
 DT Journal  
 LA English  
 AB The effects of process parameters on the mech. property and visual aspect of Y2O3 coating formed by plasma spraying were investigated. For the expts., granular Y2O3 powders were plasma sprayed using plasma gun with external powder-feeding system. When an asym. spraying was introduced by one-side powder-feeding, bigger pores and more frequent black spots were found in the region formed in opposite side from powder feeding direction, where the droplets of relatively lower temperature and lower flying velocity impinged to form coating. The optimum conditions for highest Vickers hardness were found to be different from those for white visual aspect of the Y2O3 coatings. Among the process parameters such as Ar flow rate, H2 flow rate, O2 contents, spraying distance, spraying distance was known to be most important for the formation of black spots in Y2O3 coating. The black spots formed in Y2O3 coating are thought to have relationship with the breakdown of stoichiometry between Y and O during plasma spraying. Consequently, the processing route not to form black spots in Y2O3 coating is described.

CC 57 (Ceramics)

# RETABLE

Referenced Author (RAU)	Year (RPY)	VOL (RVL)	PG (RPG)	Referenced Work (RWK)	Referenced File
=====	=====	=====	=====	=====	=====
Fauchais, P	2000	39	852	Int J Therm Sci	HCAPLUS
Jiansirisomboon, S	2003	23	961	J Euro Ceram Soc	HCAPLUS
Ross, P	1996			Taguchi Techniques f	
Sampath, S	2001	304-3	144	Materials Science an	
Vardelle, M	1996	68	1093	Pure & Appl Chem	HCAPLUS
Wuttiphphan, S	1997	293	251	Thin Solid Films	HCAPLUS

L73 ANSWER 2 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 2007:119376 HCAPLUS Full-text

DN 146:154169

TI Y2O3 thermal sprayed film coating

IN Harada, Yoshio; Teratani, Takema

PA Tocalo Co., Ltd., Japan

SO PCT Int. Appl., 30pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	-----	-----	-----	-----
PI WO 2007013184	A1	20070201	WO 2005-JP14356	20050729 <--
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU,				

*ref Application*

ZA, ZM, ZW

RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,  
IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ,  
CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH,  
GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,  
KG, KZ, MD, RU, TJ, TM

EP 1780298 A1 20070502 EP 2005-768739 20050729 <--

R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,  
IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, AL,  
BA, HR, MK, YU

US 2007026246 A1 20070201 US 2005-560522 20051213 <--

KR 2007030718 A 20070316 KR 2006-701290 20060119 <--

PRAI WO 2005-JP14356 W 20050729 <--

AB The invention relates to a member comprising a Y2O3 black thermal sprayed film formed on a surface of a base material, and a member comprising a Y2O3 black thermal sprayed film coated on a surface of a base material through an undercoat or an intermediate layer. This film has good thermal properties and good damage resistance by virtue of increased hardness. The Y2O3 black thermal sprayed film is formed by subjecting a white Y2O3 powder material to plasma spraying in a substantially oxygen-free inert gas atmospheric, or by forming a white thermal sprayed film of Y2O3, then exposing this film to an electron beam or a laser beam to heat melt at least a part of the white film, thereby smoothing the film surface and, at the same time, blackening the film.

CC 76-11 (Electric Phenomena)

Section cross-reference(s): 56, 73

ST yttrium oxide thermal plasma spray  
black film coating laser

IT Coating process

(plasma spraying; Y2O3 thermal  
sprayed film coating)

IT Coating process

(thermal spraying; Y2O3 thermal  
sprayed film coating)

IT 1314-36-9, Yttria, uses 1344-28-1, Aluminum  
oxide, uses

RL: TEM (Technical or engineered material use); USES (Uses)  
(Y2O3 thermal sprayed film coating)

IT 1314-36-9, Yttria, uses

RL: TEM (Technical or engineered material use); USES (Uses)  
(Y2O3 thermal sprayed film coating)

RN 1314-36-9 HCAPLUS

CN Yttrium oxide (Y2O3) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RETABLE

Referenced Author (RAU)	Year (RPY)	VOL (RVL)	PG (RPG)	Referenced Work (RWK)	Referenced File
The Kansai Electric Pow	2004			JP 2004149915 A	HCAPLUS
Tocalo Co Ltd	2001			EP 1156130 A1	HCAPLUS
Tocalo Co Ltd	2001			WO 2001042526 A1	
Tocalo Co Ltd	2001			JP 2001164354 A	HCAPLUS
Tocalo Co Ltd	2001			KR 2002003367 A	
Tocalo Co Ltd	2001			US 2002177001 A1	HCAPLUS
Tocalo Co Ltd	2001			TW 486758 A	HCAPLUS
Tocalo Co Ltd	2005			JP 2005256098 A	HCAPLUS

L73 ANSWER 3 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 2006:949852 HCAPLUS Full-text

DN 145:325234

TI **Black** conductive compositions, **black** electrodes, and  
 methods of forming thereof  
 IN Barker, Michael F.; Smith, Jerome David; Hayakawa, Keiichiro  
 PA E.I. Du Pont de Nemours and Company, USA  
 SO Eur. Pat. Appl., 27 pp.  
 CODEN: EPXXDW  
 DT Patent  
 LA English  
 FAN.CNT 4

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 1701212	A2	20060913	EP 2006-251263	20060309
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK, BA, HR, IS, YU				
	US 2006202174	A1	20060914	US 2006-369551	20060307
	KR 2006099443	A	20060919	KR 2006-22434	20060309
	CN 1848303	A	20061018	CN 2006-10082026	20060309
	JP 2006321976	A	20061130	JP 2006-64796	20060309 <--
PRAI	US 2005-660013P	P	20050309		
	US 2006-369551	A	20060307		

AB The invention is directed to **black** conductive composition(s), **black** electrodes made from such compns. and methods of forming such electrodes. In particular, the invention is directed to flat panel display applications, including alternating-current display panel applications. Still further, the invention is directed to composition(s) utilizing conductive metal oxides selected from an oxide of two or more elements selected from Ba, Ru, Ca, Cu, Sr, Bi, Pb, and the rare earth metals and photocrosslinkable polymers. These compns. are particularly useful in making photo-imageable **black** electrodes for flat panel display applications.

CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

ST **black** conductive compn electrode flat panel display

IT Electrodes

(**black** conductive compns., **black** electrodes for display device)

IT Films

(elec. conductive; **black** conductive compns., **black** electrodes for display device)

IT Electric conductors

(films; **black** conductive compns., **black** electrodes for display device)

IT Optical imaging devices

(flat panels; **black** conductive compns., **black** electrodes for)

IT 1303-86-2, Boron oxide, uses 1304-76-3, Bismuth oxide, uses 1305-78-8, Calcium oxide, uses 1314-13-2, Zinc oxide, uses 7429-90-5, Aluminum, uses 7631-86-9, Silica, uses 12009-17-5, Barium ruthenium oxide (BaRuO3) 12053-92-8, Copper lanthanum oxide (CuLa2O4) 12169-14-1, Ruthenium strontium oxide (RuSrO3) 12313-89-2, Calcium ruthenium oxide (CaRuO3) 25086-15-1, Carboset XPD1234 25086-89-9, Vinylacetate-vinylpyrrolidone copolymer 51183-88-1, Copper lanthanum strontium oxide (CuLaSrO4) 110778-78-4, Barium copper lanthanum oxide (BaCuLaO4) 111591-04-9, Barium copper neodymium oxide (Ba2Cu3NdO7) 112872-70-5, Barium copper yttrium oxide (Ba2Cu2YO7) 115866-34-7, Bismuth calcium copper strontium oxide (Bi2CaCu2Sr2O8) 121110-98-3, Copper neodymium strontium oxide (Cu3NdSr2O7)

RL: TEM (Technical or engineered material use); USES (Uses)

(**black** conductive compns., **black** electrodes for display device containing)

L73 ANSWER 4 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN  
 AN 2005:1023676 HCAPLUS Full-text  
 DN 143:330566  
 TI **Black-colored sprayed yttria-coated parts**  
 having good wear resistance and their manufacture  
 IN Harada, Yoshio; Teratani, Takema  
 PA Tocalo Co., Ltd., Japan  
 SO Jpn. Kokai Tokkyo Koho, 14 pp.  
 CODEN: JKXXAF  
 DT Patent  
 LA Japanese  
 FAN.CNT 1

12/13/2005

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2005256098	A	20050922	JP 2004-69925	20040312
	JP 2006118053	A	20060511	JP 2005-362629	20051216
PRAI	JP 2004-69925	A3	20040312		

AB The Y2O3 coating is formed on a substrate surface, preferably via an undercoat and an interlayer. The undercoat, having thickness 50-500 μm, may be made of Ni, Ni alloy, W, W alloy, Mo, Mo alloy, Ti, Ti alloy, Al, Al alloy, and/or Mg alloy.

IC ICM C23C0004-10  
ICS C23C0004-18

CC 57-2 (Ceramics)  
Section cross-reference(s): 56

ST **black colored sprayed yttria coated part**  
wear resistance

IT **Coating materials**  
(abrasion-resistant; **black-colored sprayed yttria-coated parts** having good wear resistance and their manufacture)

IT **Ceramic coatings**  
(**black-colored sprayed yttria-coated parts** having good wear resistance and their manufacture)

IT **Coating process**  
(**spray; black-colored sprayed yttria-coated parts** having good wear resistance and their manufacture)

IT **Aluminum alloy, base**  
**Magnesium alloy, base**  
**Molybdenum alloy, base**  
**Nickel alloy, base**  
**Titanium alloy, base**  
**Tungsten alloy, base**

RL: TEM (Technical or engineered material use); USES (Uses)  
(undercoat; **black-colored sprayed yttria-coated parts** having good wear resistance and their manufacture)

IT 1314-36-9P, Yttria, preparation  
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
(**black-colored sprayed yttria-coated parts** having good wear resistance and their manufacture)

IT 7429-90-5, Aluminum, uses 7439-98-7, Molybdenum, uses 7440-02-0, Nickel, uses 7440-32-6, Titanium, uses 7440-33-7, Tungsten, uses  
RL: TEM (Technical or engineered material use); USES (Uses)  
(undercoat; **black-colored sprayed yttria-coated parts** having good wear resistance and their manufacture)

InIDS/Fl.PR

IT 1314-36-9P, Yttria, preparation  
 RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
 (black-colored sprayed yttria-coated  
 parts having good wear resistance and their manufacture)  
 RN 1314-36-9 HCAPLUS  
 CN Yttrium oxide (Y2O3) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L73 ANSWER 5 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN  
 AN 2004:59623 HCAPLUS Full-text  
 DN 140:115405  
 TI Rare-earth oxide powders modified for dark thermal-spray  
 coating on articles  
 IN Maeda, Takao  
 PA Shin-Etsu Chemical Co., Ltd., Japan  
 SO U.S. Pat. Appl. Publ., 5 pp.  
 CODEN: USXXCO  
 DT Patent  
 LA English  
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	-----	----	-----	-----	-----
PI	US 2004013911	A1	20040122	US 2003-618679	20030715 <--
	US 6852433	B2	20050208		
	JP 2004100039	A	20040402	JP 2003-190086	20030702 <--
PRAI	JP 2002-211400	A	20020719		

AB The white rare-earth oxide powders for thermal-spray coating on metal or ceramic articles are modified with minor C, Ti, or Mo to form the coating with dark gray or black color. The modified oxide powders contain 0.1-2% C, or 1-1000 ppm of Ti or Mo. The Y2O3 powder was stirred in aqueous 30% sucrose for 10 min, filtered, dried, and fired at 1630° in flowing Ar for darkening with 1.0% of residual C. The blackened powder was used for plasma spray coating of Al-alloy substrate, forming dark coating 200 µm thick. The dark coating was stable in plasma-exposure test on Si substrate at 13.56 MHz and 1000 W in flowing CF4-O2 gas mixture

IC ICM B32B0009-00

INCL 428702000

CC 56-6 (Nonferrous Metals and Alloys)

Section cross-reference(s): 57

ST rare earth oxide powder darkening thermal spray  
 coating

IT Rare earth oxides

RL: PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)

(powder, spray coating with; white rare-earth oxide powders  
 modified for dark or black thermal-spray  
 coating)

IT Coating process

(thermal spraying, oxide, darkening of;  
 white rare-earth oxide powders modified for dark or  
 black thermal-spray coating)

IT Aluminum alloy, base

RL: PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)

(coating of; white rare-earth oxide powders modified for dark  
 or black thermal-spray coating)

IT 57-50-1, Sucrose, uses 108-95-2, Phenol, uses

RL: MOA (Modifier or additive use); USES (Uses)

(C from fired, oxide powders darkened with; white rare-earth oxide powders modified for dark or black thermal-spray coating)

IT 10241-05-1, Molybdenum pentachloride

RL: MOA (Modifier or additive use); USES (Uses)

(Mo from fired, oxide powders darkened with; white rare-earth oxide powders modified for dark or black thermal-spray coating)

IT 7705-07-9, Titanium trichloride, uses

RL: MOA (Modifier or additive use); USES (Uses)

(Ti from fired, oxide powders darkened with; white rare-earth oxide powders modified for dark or black thermal-spray coating)

IT 7439-98-7, Molybdenum, uses 7440-32-6, Titanium, uses 7440-44-0, Carbon, uses

RL: MOA (Modifier or additive use); USES (Uses)

(oxide powders with; white rare-earth oxide powders modified for dark or black thermal-spray coating)

IT 1314-36-9, Yttria, uses 1314-37-0, Ytterbium oxide

RL: TEM (Technical or engineered material use); USES (Uses)

(powder, darkening of, for spray coating; white rare-earth oxide powders modified for dark or black thermal-spray coating)

IT 1314-36-9, Yttria, uses

RL: TEM (Technical or engineered material use); USES (Uses)

(powder, darkening of, for spray coating; white rare-earth oxide powders modified for dark or black thermal-spray coating)

RN 1314-36-9 HCAPLUS

CN Yttrium oxide (Y2O3) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

# RETABLE

Referenced Author (RAU)	Year (RPY)	VOL (RVL)	PG (RPG)	Referenced Work (RWK)	Referenced File
====	====	====	====	====	====
Anon	2001			JP 2001164354 A	HCAPLUS
Harada	2002			US 20020177001 A1	
Omori	1985			US 4502983 A	HCAPLUS
Oscarsson	1999			US 5993970 A	HCAPLUS
Sperlich	2000			US 6080232 A	HCAPLUS
Swiler	2003			US 6582814 B2	HCAPLUS
Takai	2002			U.S. Appl. No. 10101	
Takai	2002			U.S. Appl. No. 10173	
Takai	2002			U.S. Appl. No. 10173	

L73 ANSWER 6 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 1998:39427 HCAPLUS Full-text

DN 128:105261

TI Thermal barrier coating experience in gas turbine engines at Pratt & Whitney

AU Bose, S.; DeMasi-Marcin, J.

CS Pratt and Whitney, East Hartford, CT, 06108, USA

SO Journal of Thermal Spray Technology (1997), 6(1), 99-104

CODEN: JTTEES; ISSN: 1059-9630

PB ASM International

DT Journal

LA English

AB Pratt & Whitney has accumulated more than three decades of experience with thermal barrier coatings (TBCs). These coatings were originally developed to



reduce surface temps. of combustors of JT8D gas turbine engines to increase the thermal fatigue life of the components. Continual improvements in design, processing, and properties of TBCs have extended their applications to other turbine components, such as vanes, vane platforms, and blades, with attendant increases in performance and component durability. Plasma-spray-based generation I (Gen I) combustor TBCs with 7 weight% yttria partially stabilized zirconia deposited by air plasma spray (APS) on an APS NiCoCrAlY bond coat continues to perform extremely well in all product line engines. Durability of this TBC has been further improved in Gen II TBCs for vans by incorporating low-pressure chamber plasma-sprayed NiCoCrAlY has a bond coat. The modification has improved TBC durability by a factor of 2.5 and altered the failure mode for a "black failure" within the bond coat to a "white failure" within the ceramic. Further improvements have been accomplished by instituting a more strain-tolerant ceramic top layer with electron beam/phys. vapor deposition (EB-PVD) processing. This Gen III TBC has demonstrated exceptional performance on rotating airfoils in high-thrust-rated engines, improving blade durability by three times through elimination of blade creep, fracture, and rumpling of metallic coatings used for oxidation protection of the airfoil surfaces. A TBC durability model for plasma-sprayed as well as EB-PVD systems is proposed that involves the accumulation of compressive stresses during cyclic thermal exposure. The model attempts to correlate failure of the various TBCs with elements of their structure and its degradation with thermocyclic exposure.

CC 57-2 (Ceramics)  
Section cross-reference(s): 56

IT Coating process  
(plasma spraying; thermal barrier coating  
experience in gas turbine engines at Pratt & Whitney and durability  
model)

IT Thermal barrier coatings  
Thermal cycling  
Thermal fatigue  
Turbines  
(thermal barrier coating experience in gas turbine engines at Pratt &  
Whitney and durability model)

IT 1314-23-4P, Zirconia, preparation  
RL: DEV (Device component use); IMF (Industrial manufacture); PRP  
(Properties); PREP (Preparation); USES (Uses)  
(yttria-stabilized, coatings; thermal barrier coating  
experience in gas turbine engines at Pratt & Whitney and durability  
model)

IT 1314-36-9P, Yttria, preparation  
RL: DEV (Device component use); IMF (Industrial manufacture); PRP  
(Properties); PREP (Preparation); USES (Uses)  
(zirconia stabilized by, coatings; thermal barrier coating experience  
in gas turbine engines at Pratt & Whitney and durability model)

IT 1314-36-9P, Yttria, preparation  
RL: DEV (Device component use); IMF (Industrial manufacture); PRP  
(Properties); PREP (Preparation); USES (Uses)  
(zirconia stabilized by, coatings; thermal barrier coating experience  
in gas turbine engines at Pratt & Whitney and durability model)

RN 1314-36-9 HCAPLUS

CN Yttrium oxide (Y2O3) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

# RETABLE

Referenced Author (RAU)	Year (RPY)	VOL (RVL)	PG (RPG)	Referenced Work (RWK)	Referenced File
Demasi, J	1989			Thermal Barrier Coat	

Meier, S	1994	116	250	J Eng Gas Turbines P	HCAPLUS
Miller, R	1982	95	265	Thin Solid Films	HCAPLUS

L73 ANSWER 7 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 1996:751071 HCAPLUS Full-text

DN 126:149486

TI Compatibility of yttria (Y2O3) with liquid lithium

AU Terai, T.; Yoneoka, T.; Tanaka, H.; Suzuki, A.; Tanaka, S.; Nakamichi, M.; Kawamura, H.; Miyajima, K.; Harada, Y.

CS Department of Quantum Engineering and Systems Science, University of Tokyo, 7-3-1, Hongo, Tokyo, 113, Japan

SO Journal of Nuclear Materials (1996), 233-237 (Pt. B), 1421-1426<sup>0</sup>

CODEN: JNUMAM; ISSN: 0022-3115

PB Elsevier

DT Journal

LA English

AB Compatibility of sintered specimens and plasma sprayed coating specimens of Y2O3 with liquid Li was tested at 773 K. No configuration change was observed for the sintered specimens with a slight increase of thickness for 1419 h. Li-Y complex oxide (LiYO2) was formed on the surface, and the inner part changed to gray or black nonstoichiometric Y2O3-x with lower elec. resistivity. The plasma sprayed coating specimens were severely attacked by liquid Li with or without applied elec. field. Li penetrated into the coating layer through small cracks and reacted on Y2O3 to form LiYO2, which has a different d. from Y2O3 and is more brittle than Y2O3. Y2O3 has a possibility as a ceramic coating material for liquid blankets if it can be made into a dense coating on the surface of piping materials.

CC 71-2 (Nuclear Technology)

ST compatibility yttria liq lithium liq blanket; fusion reactor blanket yttria compatibility

IT Fusion reactor blankets

(compatibility of yttria (Y2O3) with liquid lithium)

IT 12169-03-8, Lithium yttrium oxide (LiYO2)

RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)

(compatibility of yttria (Y2O3) with liquid lithium)

IT 1314-36-9, Yttrium oxide (Y2O3),

uses

RL: NUU (Other use, unclassified); RCT (Reactant); RACT (Reactant or reagent); USES (Uses)

(compatibility of yttria (Y2O3) with liquid lithium)

IT 7439-93-2, Lithium, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(compatibility of yttria (Y2O3) with liquid lithium)

IT 1314-36-9, Yttrium oxide (Y2O3),

uses

RL: NUU (Other use, unclassified); RCT (Reactant); RACT (Reactant or reagent); USES (Uses)

(compatibility of yttria (Y2O3) with liquid lithium)

RN 1314-36-9 HCAPLUS

CN Yttrium oxide (Y2O3) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RETABLE

Referenced Author (RAU)	Year (RPY)	VOL (RVL)	PG (RPG)	Referenced Work (RWK)	Referenced File
Bauer, A	1974	1930		Battelle Mem Inst Re	
Borgstedt, H	1995	212-2	1501	J Nucl Mater	
Cairns, E	1971			ANL-7888	
Glasbrenner, H	1995	212-2	1561	J Nucl Mater	

Hartmanova, M	1989	36	137	Solid State Ionics	HCAPLUS
Hollenberg, G	1995	28	190	Fus Eng Des	HCAPLUS
Malang, S	1995	27	570	Fus Eng Des	HCAPLUS
Nakamichi, M	1995		1217	Proc SOFT-18	HCAPLUS
Perujo, A	1995	28	252	Fus Eng Des	HCAPLUS
Roth, R	1987	6	92	Phase Diagrams for C	
Tallen, N	1966	49	401	J Am Ceram Soc	
Terai, T	1995		1329	Proc SOFT-18	HCAPLUS
Tominetti, S	1990	176-1	672	J Nucl Mater	HCAPLUS
Yoneoka, T	1996	60	1	J Jpn Inst Metals	HCAPLUS
Yoneoka, T				J Jpn Inst Metals, t	

L73 ANSWER 8 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 1993:130509 HCAPLUS Full-text

DN 118:130509

TI Carbonyl metal-containing, diffuse, optically black plasma-sprayed coatings

IN Marousek, Michael E.; Nagle, Dennis C.; Shepard, Donald F.

PA Martin-Marietta Corp., USA

SO U.S., 4 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 5176964	A	19930105	US 1991-685479	19910412 <--
PRAI	US 1991-685479		19910412		

AB The coatings have absorptivity .gtorsim.0.92 and emissivity .gtorsim.0.85, are obtained by plasma spraying substrates with a powder composition comprising approx. 20-50 weight% carbonyl metal and balance ceramic oxide. The substrates, preferably Ti-6Al-4V, may first be coated with a bonding layer by plasma spraying a NiCrAl alloy. The ceramic oxide is Y2O3-stabilized ZrO2, and the carbonyl metal is carbonyl iron. These coatings are stable in space environment, and have high absorptivity for solar energy and high IR emittance.

IC ICM B22F0003-00

ICS B32B0009-04

INCL 428552000

CC 57-2 (Ceramics)

Section cross-reference(s): 56

ST UV absorbing IR emitting coating; carbonyl iron powder zirconia coating; yttria stabilizer zirconia iron powder; plasma spraying coating substrate; titanium aluminum vanadium alloy substrate; bonding interlayer coating substrate; nickel chromium aluminum alloy interlayer; optical black coating space vehicle

IT Metals, uses

RL: USES (Uses)

(powdered, optically black coating materials containing ceramic oxide and, for space vehicles, for high absorptivity for solar energy and high IR emittance)

IT Coating materials

(optically black, coating with, of substrates, for space vehicles, for high absorptivity for solar energy and high IR emittance)

IT Coating process

(plasma spraying, of substrates, with optically black coating materials, for high absorptivity for solar energy and high IR emittance, for space vehicles)

IT Ceramic materials and wares  
 (powdered, oxides, optically **black** coating materials containing metal powdered and, for space vehicles, for high absorptivity for solar energy and high IR emittance)

IT 12743-70-3, Ti-6Al-4V  
 RL: USES (Uses)  
 (coating of, by plasma **spraying**, with ceramic oxide- and metal powder-containing optically **black** coating materials, for space vehicles, for high absorptivity for solar energy and high IR emittance)

IT 55467-16-8, Aluminum, chromium, nickel base  
 RL: USES (Uses)  
 (powdered, coating with, by plasma **spraying**, of substrates for bonding interlayer, in, optically **black** coating formation on space vehicles, for high absorptivity for solar energy and high IR emittance)

IT 7439-89-6P, Iron, preparation  
 RL: PREP (Preparation)  
 (powdered, optically **black** coating materials containing ceramic oxide and, for space vehicles, for high absorptivity for solar energy and high IR emittance)

IT 1314-23-4, Zirconia, uses  
 RL: USES (Uses)  
 (yttria-stabilized, optically **black** coating materials containing metal powdered and, for space vehicles, for high absorptivity for solar energy and high IR emittance)

IT 1314-36-9, Yttria, uses  
 RL: USES (Uses)  
 (zirconia stabilized with, optically **black** coating materials containing metal powdered and, for space vehicles, for high absorptivity for solar energy and high IR emittance)

IT 1314-36-9, Yttria, uses  
 RL: USES (Uses)  
 (zirconia stabilized with, optically **black** coating materials containing metal powdered and, for space vehicles, for high absorptivity for solar energy and high IR emittance)

RN 1314-36-9 HCAPLUS  
 CN Yttrium oxide (Y2O3) (CA INDEX NAME)


\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L73 ANSWER 9 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN  
 AN 1986:114196 HCAPLUS Full-text  
 DN 104:114196  
 TI Plasma **sprayed** aluminum-yttria containing  
 (1) iron-chromium and (2) nickel-chromium coatings with excellent corrosion resistance and **sprayability** factors for use in energy conversion systems

AU Harrington, John H.; Rangaswamy, Subramaniam  
 CS METCO, Westbury, NY, USA  
 SO Coat. Bimet. Aggressive Environ., Conf. Proc. (1985), Meeting Date 1984, 171-8. Editor(s): Sisson, Richard D., Jr. Publisher: ASM, Metals Park, Ohio.  
 CODEN: 54WHAK

DT Conference  
 LA English  
 AB Fe-27Cr-9Al-2Y2O3 [100458-57-9] and Ni-18Cr-9Al-2Y2O3 [100458-56-8] coatings formed by plasma **spraying** of composite powders exhibited excellent corrosion

resistance and improved **sprayability** and bonding over state-of-the-art coatings. The powders were formulated to utilize the Al-Y2O3 alloying reactions in the plasma flame to provide superior corrosion resistance and **spraying** properties. The coatings were evaluated for high-temperature corrosion resistance to Na, S, Cl compds., and oxidation by using industrial screening tests. The Fe-base coatings had superior corrosion resistance in reducing atms. involving a dry char of Na2S-Na2CO3-NaCl at 850°F to simulate a **black liquor recovery boiler**. The Ni-base coating had superior corrosion resistance in oxidizing atms. in a laboratory molten salt crucible test involving Na2SO4-NaCl at 1380°F. The improved **sprayability** factors resulting from the composites are higher bond strengths, greater thickness limits, better bend ductility, and less **spray** technique dependence for on-site **spraying** applications.

- CC 56-6 (Nonferrous Metals and Alloys)  
 ST **nickel alloy coating boiler corrosion; iron alloy coating boiler corrosion; chromium addn plasma **sprayed** coating; aluminum addn plasma **sprayed** coating; yttria addn plasma **sprayed** coating**  
 IT Boilers  
     (coatings for, plasma-sprayed corrosion-resistant)  
 IT **Coating materials**  
     (iron and **nickel** alloys, plasma-sprayed corrosion-resistant, for boilers)  
 IT 100458-56-8 100458-57-9  
 RL: USES (Uses)  
     (coatings of, plasma-sprayed corrosion-resistant, for boilers)
- L73 ANSWER 10 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN  
 AN 1981:200209 HCAPLUS Full-text  
 DN 94:200209  
 TI Properties of **black yttrium oxide sintered bodies**  
 AU Tsukuda, Y.  
 CS Cent. Res. Lab., Hitachi Ltd., Kokubunji, 185, Japan  
 SO Materials Research Bulletin (1981), 16(4), 453-9  
 CODEN: MRBUAC; ISSN: 0025-5408  
 DT Journal  
 LA English  
 AB  **Black Y2O3 pieces were obtained by heating in a reducing atmospheric, and they have some properties other types of Y2O3 do not have. Hardness, transmittance, and thermoluminescence of black Y2O3 sintered pieces were investigated. The Knoop hardness nos. of the black Y2O3 pieces vary from 615 to 804 kg/mm2, and the average hardness number is 699 kg/mm2, which is nearly equal to that of a colorless piece. In-line transmittances of the dark Y2O3 pieces in the 0.2 .apprx. 11 μm wavelength are lower than those of colorless pieces. The trap level of the black Y2O3 pieces is 1.22 eV.**
- CC 73-3 (Spectra by Absorption, Emission, Reflection, or Magnetic Resonance, and Other Optical Properties)  
 ST **yttrium oxide black optical property; Knoop hardness black yttrium oxide; thermoluminescence black yttrium oxide; luminescence black yttrium oxide; IR visible black yttrium oxide**  
 IT Trapping and Traps  
     (in yttrium oxide black centered bodies)  
 IT Infrared spectra  
 Luminescence, thermo-  
 Ultraviolet and visible spectra  
     (of yttrium oxide black centered bodies)

IT 1314-36-9, properties  
RL: PRP (Properties)  
(Knoop hardness nos. and optical properties of black centered  
bodies of)  
IT 1314-36-9, properties  
RL: PRP (Properties)  
(Knoop hardness nos. and optical properties of black centered  
bodies of)  
RN 1314-36-9 HCAPLUS  
CN Yttrium oxide (Y2O3) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L73 ANSWER 11 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN  
AN 1980:576190 HCAPLUS Full-text  
DN 93:176190  
TI Emissivity of **sprayed** layers of **yttrium oxide**  
and solid solutions of zirconium dioxide  
AU Zhorov, G. A.; Zakharov, B. M.  
CS Moscow, USSR  
SO Teplofizika Vysokikh Temperatur (1980), 18(4), 745-51  
CODEN: TVYTAP; ISSN: 0040-3644  
DT Journal  
LA Russian  
AB The normal integral and semispherical degree of **blackness** was determined at  
400-2500 K of layers of Y2O3, Zr, ZrO2-CaO, and ZrO2-CaO-SiO2 solid solns.  
The effect of vacuum depth on the emissivity of **spray** deposited layers at high  
temps. was studied.  
CC 73-2 (Spectra by Absorption, Emission, Reflection, or Magnetic Resonance,  
and Other Optical Properties)  
ST emissivity **yttria** zirconia solid soln  
IT Emissivity  
(of **yttria** and zirconia solid solns. **sprayed**  
layers)  
IT 1305-78-8D, solid solns. with silica and zirconium dioxide 1314-23-4D,  
solid solns. with calcium oxide and silicon dioxide 1314-36-9,  
properties 7440-67-7, properties 7631-86-9D, solid solns. with calcium  
oxide and zirconium oxide  
RL: PRP (Properties)  
(emissivity of plasma **sprayed** layers of)  
IT 1314-36-9, properties  
RL: PRP (Properties)  
(emissivity of plasma **sprayed** layers of)  
RN 1314-36-9 HCAPLUS  
CN Yttrium oxide (Y2O3) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L73 ANSWER 12 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN  
AN 1979:96568 HCAPLUS Full-text  
DN 90:96568  
TI High thermal emittance coating for x-ray targets  
IN Hueschen, Robert E.; Jens, Richard A.  
PA General Electric Co., USA  
SO U.S., 7 pp.  
CODEN: USXXAM  
DT Patent  
LA English  
FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	US 4132916	A	19790102	US 1977-769067	19770216
	DE 2805154	A1	19781123	DE 1978-2805154	19780208
	FR 2381834	A1	19780922	FR 1978-3942	19780213 <--
	FR 2381834	B1	19830805		
	JP 53108796	A	19780921	JP 1978-16005	19780216
PRAI	US 1977-769067	A	19770216		

AB A high thermal emittance coating for an x-ray tube anode target comprises a high m.p. oxide or a mixture of such oxides plus TiO<sub>2</sub> and a stabilizer of CaO or Y<sub>2</sub>O<sub>3</sub>. The product results from heating at  $\leq 10^{-5}$  torr and 1650-1900° a mixture of TiO<sub>2</sub> 2.5-20, an oxide (ZrO<sub>2</sub>, HfO, MgO, CeO<sub>2</sub>, La<sub>2</sub>O<sub>3</sub>, or SrO) 70-93.5 weight %, and a stabilizer of CaO or Y<sub>2</sub>O<sub>3</sub> balance. The coating is fused and bonded tightly to the target to minimize flaking off. For example, specific oxide coating compns. which produce a black fused coating with thermal emittance values of 0.92-0.94 include ZrO<sub>2</sub> 76, CaO 4, and TiO<sub>2</sub> 20 weight %; and ZrO<sub>2</sub> 87.88, CaO 4.62, and TiO<sub>2</sub> 7.5 weight %.

IC H01J0035-08

INCL 313330000

CC 76-14 (Electric Phenomena)

IT Coating materials

(oxide fused, with high thermal emittance, for x-ray tube anode target)

L73 ANSWER 13 OF 13 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 1976:168583 HCAPLUS Full-text

DN 84:168583

TI Studies on sintering yttrium oxide

AU Tsukuda, Yasuo; Muta, Akinori

CS Cent. Res. Lab., Hitachi Ltd., Kokubunji, Japan

SO Yogyo Kyokaishi (1976), 84(3), 141-7

CODEN: YGKSA4; ISSN: 0009-0255

DT Journal

LA Japanese

AB Y<sub>2</sub>O<sub>3</sub> [1314-36-9] is chemical stable and one of the hopeful refractory materials. It is, however, difficult to obtain dense Y<sub>2</sub>O<sub>3</sub> pieces using an ordinary sintering method without any additives. The effect of sintering conditions, such as temperature, time, and atmospheric on sinterability of Y<sub>2</sub>O<sub>3</sub> were investigated. Total impurity of Y<sub>2</sub>O<sub>3</sub> was <0.1% and the average grain size of the powder was 4-6  $\mu$ . The compacting pressure was 0.17-17 tons/cm<sup>2</sup>. The green compacts were heated at 1700-2120° for 30-10/ min in a H or Ar atmospheric. When the d. of green compacts was 2.0-2.7 g/cm<sup>3</sup>, the d. of sintered pieces in a H atmospheric increased with that of green compacts. The sintered pieces became black from the inside. In-line transmittance of the sintered pieces in a H increased with the grain size. Activation energy for the grain growth was 90 kcal/mole. The pieces sintered in an Ar atmospheric had a brown or dark surface layer where Mo was detected. A translucent Y<sub>2</sub>O<sub>3</sub> piece was obtained, which has in-line transmittance of 70% at 0.76 mm thickness in the visible range.

CC 57-5 (Ceramics)

ST yttria refractory sintering

IT Sintering

(of yttrium oxide refractory)

IT Refractories

(yttrium oxide, sintering of translucent)

IT 1314-36-9

RL: USES (Uses)

(refractories, sintering of translucent)

IT 1314-36-9

RL: USES (Uses)

(refractories, sintering of translucent)

RN 1314-36-9 HCAPLUS

CN Yttrium oxide (Y2O3) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

=> => fil wpix

FILE 'WPIX' ENTERED AT 16:03:09 ON 19 SEP 2007

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MOST RECENT THOMSON SCIENTIFIC UPDATE: 200759 <200759/DW>  
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>>> Indian patent publication number format enhanced in DWPI - see NEWS <<<

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'BI ABEX' IS DEFAULT SEARCH FIELD FOR 'WPIX' FILE

=> d bib ab tech abex tot l102

L102 ANSWER 1 OF 4 WPIX COPYRIGHT 2007 THE THOMSON CORP on STN

AN 2007-271761 [26] WPIX Full-text

DNC C2007-099112 [26]

DNN N2007-202130 [26]

TI Yttrium oxide sprayed film coated component used for  
heater pipe, consists of black sprayed coating of  
yttrium oxide provided on surface of base material

DC L02; M13; P42; P73

IN HARADA Y; TERATANI T

PA (TOCA-N) TOCALO CO LTD

CYC 110

PIA WO 2007013184 A1 20070201 (200726)\* JA 30[1]

US 20070026246 A1 20070201 (200726) EN

EP 1780298 A1 20070502 (200731) EN

KR 2007030718 A 20070316 (200755) KO

ADT WO 2007013184 A1 WO 2005-JP14356 20050729; EP 1780298 A1 EP

2005-768739 20050729; US 20070026246 A1 WO 2005-JP14356

20050729; EP 1780298 A1 WO 2005-JP14356 20050729; US

20070026246 A1 US 2005-560522 20051213; KR 2007030718 A WO

2005-JP14356 20050729; KR 2007030718 A KR 2006-701290 20060119

FDT EP 1780298 A1 Based on WO 2007013184 A; KR 2007030718 A Based on  
WO 2007013184 A

PRAI WO 2005-JP14356 20050729

AB WO 2007013184 A1 UPAB: 20070423



NOVELTY - The yttrium oxide sprayed film coated component consists of black sprayed coating of yttrium oxide provided on the surface of a base material.  
DETAILED DESCRIPTION - An INDEPENDENT CLAIM is included for manufacture of the yttrium oxide sprayed film coated component.

USE - For heater pipe and for infrared radiant section material.

ADVANTAGE - The yttrium oxide sprayed film coated component has excellent thermal radiation property and damage resistance. Since the spray coating of yttria becomes black, the contamination of the film is reduced and repeated washing of the film is eliminated. The yttria film has high hardness and improved wear resistance.

#### TECH

INORGANIC CHEMISTRY - Preferred Composition: The coated component further comprises undercoat containing metallic film, and topcoat. The metallic film contains nickel, tungsten, aluminum, molybdenum, titanium and/or magnesium, or their alloys. An intermediate layer containing alumina and/or yttria is further provided between the undercoat and topcoat.

L102 ANSWER 2 OF 4 WPIX COPYRIGHT 2007

THE THOMSON CORP on STN

AN 2006-336337 [35] WPIX Full-text

CR 2005-668927

DNC C2006-111154 [35]

DNN N2006-284610 [35]

TI Base material for semiconductor manufacturing apparatus coated with yttrium oxide by laser beam processing or electron beam processing black spray coating

DC L03; U11; X25

IN HARADA Y; TERATANI T

PA (TOCA-N) TOCALO CO LTD

CYC 1

PIA JP 2006118053 A 20060511 (200635)\* JA 15[1]

ADT JP 2006118053 A Div Ex JP 2004-69925 20040312; JP 2006118053 A  
JP 2005-362629 20051216.

PRAI JP 2005-362629 20051216

JP 2004-69925 20040312

AB JP 2006118053 A UPAB: 20060602

NOVELTY - The base material is coated with yttrium oxide by laser beam processing or electron beam processing black spray coating.

DETAILED DESCRIPTION - An undercoat which consists of metal film is provided between base material and yttrium oxide coating. The metallic film is chosen from nickel, tungsten, molybdenum, titanium, aluminum, magnesium or their alloy.

USE - For semiconductor manufacturing apparatus, also in baffle plate, focus ring, insulator ring, shielding ring, bellows cover, electrode and metal melt crucible.

ADVANTAGE - The base material has excellent thermal radiation property, damage resistance and contamination resistance.

L102 ANSWER 3 OF 4 WPIX COPYRIGHT 2007

THE THOMSON CORP on STN

AN 2005-668927 [69] WPIX Full-text

CR 2006-336337

DNC C2005-203346 [69]

DNN N2005-548045 [69]

TI Yttrium-oxide spray-coating covered component for baffle plate, has surface of base material covered with black -spray coating of yttrium oxide

DC M13; V05

IN HARADA Y; TERATANI T

PA (TOCA-N) TOCALO CO LTD

CYC 1

PIA JP 2005256098 A 20050922 (200569)\* JA 14[1]

ADT JP 2005256098 A JP 2004-69925 20040312

PRAI JP 2004-69925 20040312

AB JP 2005256098 A UPAB: 20051223

NOVELTY - The yttrium-oxide spray-coating covered component has a surface of a base material covered with black -spray coating of yttrium oxide.

DETAILED DESCRIPTION - An INDEPENDENT CLAIM is included for manufacture of the yttrium-oxide spray-coating covered component.

USE - For baffle plate, focus rings, insulator rings, shielding rings, bellow covers and electrodes.

ADVANTAGE - The yttrium oxide spray-coating covered component has excellent thermal-radiation property, thermal-shock resistance, damage resistance, adhesivity, glossiness, plasma-proof erosion property, alkali resistance, and working efficiency such as plasma-etching processing, and is manufactured economically with excellent production efficiency, without generation of crack.

TECH

INORGANIC CHEMISTRY - Preferred Component: The yttrium-oxide spray-coating covered component has a metallic film containing metal such as nickel, tungsten, molybdenum, titanium, aluminum, magnesium and/or their alloys as an undercoat, and an intermediate layer formed using solid solution containing alumina and/or yttrium oxide.

ORGANIC CHEMISTRY - Preferred Component: The yttrium-oxide spray-coating covered component has a metallic film containing metal such as nickel, tungsten, molybdenum, titanium, aluminum, magnesium and/or their alloys as an undercoat, and an intermediate layer formed using solid solution containing alumina and/or yttrium oxide.

L102 ANSWER 4 OF 4 WPIX COPYRIGHT 2007

THE THOMSON CORP on STN

AN 1990-299743 [40] WPIX Full-text

DNC C1990-129434 [21]

TI Metal oxide coating for ceramic oxide fibres - produced by reactive CVD in flow chamber to give material suitable for reinforcing metals or alloys

DC L02; M22

IN BOUIX J; HILLEL R

PA (CNRS-C) CNRS CENT NAT RECH SCI

CYC 1

PIA FR 2643088 A 19900817 (199040)\* FR

&lt;--

ADT FR 2643088 A FR 1989-2369 19890216

PRAI FR 1989-2369 19890216

AB FR 2643088 A UPAB: 20050501

A coating suitable for ceramic oxides in fibres or in bulk is claimed in which the coating is a polyvalent metallic compound such as titanium oxide. The method claimed produces the 0.1 to 1.0 micron coating by reactive CVD in H<sub>2</sub> with the metal halide in the vapour phase at greater than 500 deg.C. The metal halide is chosen from transition metals and semiconductors; the substrate is one of silica, alumina, mullite, magnesia, zirconia or yttria, and can be mono or multi filament, a mesh or in bulk. The coating claimed consists of two layers; the outer layer of e.g. TiO<sub>x</sub> where x is between 0.35 and 0.5, and the inner layer of a complex phase consisting of, for example, in the case of mullite, Ti, Al, Si, and oxygen. The substrate as filament is moved through the reactor at constant speed preferably in an atmosphere of HCl and Ar with the metal chloride, to produce an even blackish deposit.

ADVANTAGE - The fibre characteristics of tensile strength and brittleness are retained after coating. The deposit is adherent, continuous, and homogeneous; the reaction conditions can be altered to give slightly different products with good control. It is chemically compatible with many metals and matrices

and is especially useful for coating Al<sub>2</sub>O<sub>3</sub> fibres to reinforce Al, avoiding the non-wetting behaviour of Al<sub>2</sub>O<sub>3</sub> and the peeling of SiO<sub>2</sub>. Magnesium-containing alloys can now be reinforced with the coated fibres as the reactivity of the alloy with the fibre is eliminated. The fibres have the same Young's modulus before and after treatment. The breaking stress of the coated fibres of 1750 MPa is only slightly below the value for untreated fibres of 2000 MPa. @ (18pp Dwg.No.0/2)

=> => d his

(FILE 'HOME' ENTERED AT 14:52:01 ON 19 SEP 2007)  
SET COST OFF

FILE 'REGISTRY' ENTERED AT 14:52:36 ON 19 SEP 2007

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E YTTRIUM OXIDE/CN
L1      2 S E3,E37
L2      1 S L1 AND 1/NC
L3      1 S L1 NOT L2
L4      43050 S (Y/ELS OR ?YTTRI?/CNS OR 7440-65-5/CRN) AND (O/ELS OR ?OXID?/
L5      94 S L4 AND 2/ELC.SUB
L6      80 S L5 NOT CCS/CI
L7      78 S L6 NOT L1
L8      71 S L7 NOT (PMS OR RIS)/CI
L9      7 S L8 NOT TIS/CI
        SEL RN 6 7
L10     2 S E1-E2
L11     64 S L8 NOT L9
L12     67 S L3,L10,L11

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FILE 'HCAPLUS' ENTERED AT 14:59:04 ON 19 SEP 2007

L13 43107 S L2

FILE 'HCAPLUS' ENTERED AT 14:59:19 ON 19 SEP 2007

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L14     113403 S Y2O3 OR YTTRIUM() (OXIDE OR SESQUIOXIDE OR TRIOXIDE OR 3 OXIDE
L15     114544 S L13,L14
L16     43362 S L12
L17     114625 S L15,L16
L18     725 S L17 AND ?BLACK?
L19     464 S L18 AND L13
L20     465 S L18 AND L16
L21     465 S L19,L20
L22     1 S US20070026246/PN OR (US2005-560522# OR WO2005-JP14356)/AP,PRN
        E HARADA/AU
        E HARADA Y/AU
L23     481 S E3,E4
        E HARADA YO/AU
L24     6 S E31
        E HARADA NAME/AU
L25     70 S E4
        E YOSHIO/AU
L26     3 S E3
L27     4 S E17
        E TERATANI/AU
        E TERATANI T/AU
L28     43 S E3,E7
        E TERATANI NAME/AU
        E TAKEMA/AU

```

E TOCALO/AP,CS  
 E TOCALO/PA,CS  
 L29 340 S E3-E25  
 E TOCALO/CO  
 L30 338 S E4-E7  
 E E4+ALL  
 L31 378 S E2+RT  
 E TOYO CALORIZING/PA,CS  
 L32 17 S E5-E20  
 E NIPPON COATING/PA,CS  
 L33 24 S E5-E20  
 E JAPAN COATING/PA,CS  
 L34 8 S E5-E9  
 E HARADA/AU  
 E HARADA Y/AU  
 L35 481 S E3,E4  
 E HARADA YO/AU  
 L36 493 S E51  
 L37 73 S L17 AND L22-L36  
 L38 3 S L37 AND ?BLACK?  
 E COATING PROCESS/CT  
 L39 20 S L18 AND E3-E104  
 E E3+ALL  
 L40 28 S L18 AND E9+OLD,NT  
 E E40  
 L41 20 S L18 AND E3  
 E E3+ALL  
 L42 20 S L18 AND E2+OLD  
 L43 31 S L18 AND E2+NT  
 L44 14 S L18 AND B05D/IPC,IC,ICM,ICS  
 L45 13 S L18 AND C03C/IPC,IC,ICM,ICS  
 L46 14 S L18 AND C23C/IPC,IC,ICM,ICS  
 L47 41 S L18 AND COAT?/CW,CT  
 L48 21 S L18 AND COAT?/SC,SX  
 L49 26 S L18 AND ?SPRAY?  
 L50 15 S L49 AND L39-L48  
 L51 12 S L21 AND L50  
 L52 57 S L21 AND L39-L48  
 L53 12 S L52 AND ?SPRAY?  
 L54 12 S L51,L53  
 SEL AN 1 2 4 7 9 11 12  
 L55 5 S L54 NOT E1-E14  
 L56 45 S L52 NOT L54  
 L57 75 S L39-L52 NOT L54,L55  
 L58 30 S L57 NOT L56  
 SEL AN 1 4 16 26 27 28  
 L59 6 S L58 AND E15-E25  
 L60 260 S L17 AND ?DARK?  
 L61 91 S L60 AND L13,L16  
 L62 88 S L61 NOT L39-L59  
 SEL AN 75 79  
 L63 2 S L62 AND E26-E29  
 L64 13 S L38,L55,L59,L63  
 L65 13 S L64 AND L13-L64  
 L66 8 S L65 AND (AL OR NI OR W OR MO OR TI OR MG OR NICKEL OR TUNGSTE  
 L67 13 S L65,L66  
 E LASER BEAM IRRADIATION/CT  
 E ELECTRON BEAM IRRADIATION/CT  
 E E3+ALL  
 L68 5696 S E2

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      E LASER BEAM/CT
      E E6+ALL
      E E2+ALL
L69    126362 S E10+OLD,NT
L70    763778 S E7+NT
L71      0 S L67 AND L68-L70
L72      2 S L67 AND (IR OR ?RADIAT?)
L73     13 S L67,L72
      SEL HIT RN

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FILE 'REGISTRY' ENTERED AT 15:45:31 ON 19 SEP 2007

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L74      1 S E1
L75      1 S L74 AND L1-L12

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FILE 'REGISTRY' ENTERED AT 15:45:52 ON 19 SEP 2007

FILE 'HCAPLUS' ENTERED AT 15:45:58 ON 19 SEP 2007

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L76     2790 S L2(L) PREP+NT/RL
L77     2804 S L12(L) PREP+NT/RL
L78     2804 S L76,L77
L79      23 S L78 AND ?BLACK?
L80      2 S L78 AND ?DARK?
L81     24 S L79,L80
L82     22 S L81 NOT L73

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FILE 'WPIX' ENTERED AT 15:50:04 ON 19 SEP 2007

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L83      98 S (L02-G01C OR L02-G01C1)/MC
L84     12035 S L14
      E YTTRIUM OXIDE/CN
L85      1 S E3,E34
L86      794 S R03343/DCN
L87     12362 S L83,L84,L86
L88      433 S L87 AND B05D/IPC,IC,ICM,ICS
L89      47 S L87 AND B05D001-08/IPC,IC,ICM,ICS
L90      20 S L87 AND B05D001-02/IPC,IC,ICM,ICS
L91     450 S L87 AND C23C004/IPC,IC,ICM,ICS
L92     291 S L87 AND C23C004-10/IPC,IC,ICM,ICS
L93      79 S L87 AND L02-A06/MC
L94     199 S L87 AND ?BLACK?
L95      52 S L87 AND ?DARK?
L96     232 S L94,L95 AND (PD<=20050729 OR PRD<=20050729 OR AD<=20050729)
L97      17 S L96 AND L88-L93
L98      16 S L96 AND C03C/IPC,IC,ICM,ICS
L99      18 S L96 AND B32B/IPC,IC,ICM,ICS
L100     22 S L96 AND (L02-J OR L02-J01 OR M13-H04? OR M13-M OR L02-G OR L0
L101     65 S L97-L100
      SEL AN 1 5 10 54
L102      4 S L101 AND E1-E4

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FILE 'WPIX' ENTERED AT 16:03:09 ON 19 SEP 2007

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